



## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

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<b>(21) International Application Number:</b> PCT/US99/25592 <b>(22) International Filing Date:</b> 29 October 1999 (29.10.1999) <b>(30) Priority Data:</b> 09/183,422 30 October 1998 (30.10.1998) US <b>(60) Parent Application or Grant</b> THE CURATORS OF THE UNIVERSITY OF MISSOURI [ ]; O. PHILLIPS, Jeffrey, O. [ ]; O. MAHONEY, Joseph, A. ; O.		<b>Published</b>
<b>(54) Title: OMEPRAZOLE SOLUTION AND METHOD OF USING SAME</b> <b>(54) Titre: SOLUTION D'OMEPRAZOLE ET METHODE D'UTILISATION ASSOCIEE</b>  <b>(57) Abstract</b> <p>A method of treating gastric acid disorders by administering to a patient a pharmaceutical composition including a proton pump inhibitor in a pharmaceutically acceptable carrier including a bicarbonate salt of a Group IA metal where the administering step consists of a single dosage form without requiring further administering of the bicarbonate salt of the Group IA metal. A pharmaceutical composition includes a dry formulation of a proton pump inhibitor in a pharmaceutically acceptable carrier including a bicarbonate salt of a Group IA metal. A pharmaceutical composition for making a dry formulation of a proton pump inhibitor which includes a proton pump inhibitor and a bicarbonate salt of a Group IA metal in a form for convenient storage, whereby when the composition is in a dry formulation which is suitable for enteral administration.</p> <b>(57) Abrégé</b> <p>La présente invention concerne une méthode de traitement d'acidité gastrique par administration, à un patient, d'une composition pharmaceutique contenant un inhibiteur de pompe à proton dans un support acceptable sur le plan pharmaceutique, comprenant un sel bicarbonaté d'un métal du groupe IA, qui consiste à administrer une forme à dosage unique qui ne nécessite aucune administration ultérieure de sel bicarbonaté de métal du groupe IA. Une composition pharmaceutique comprend une formulation sèche d'inhibiteur de pompe à proton dans un support acceptable sur le plan pharmaceutique, comprenant un sel bicarbonaté d'un métal du groupe IA. L'invention concerne aussi une composition pharmaceutique destinée à fabriquer une formulation sèche d'un inhibiteur de proton qui comprend un inhibiteur de pompe à proton et un sel bicarbonaté d'un métal du groupe IA sous une forme adaptée au stockage, la formulation sèche de la composition convenant à l'administration par voie entérale.</p>		

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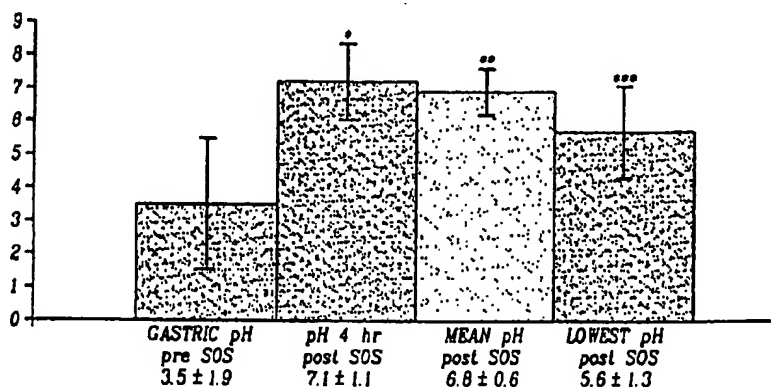
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(54) Title: OMEPRAZOLE SOLUTION AND METHOD OF USING SAME



(57) Abstract

A method of treating gastric acid disorders by administering to a patient a pharmaceutical composition including a proton pump inhibitor in a pharmaceutically acceptable carrier including a bicarbonate salt of a Group IA metal where the administering step consists of a single dosage form without requiring further administering of the bicarbonate salt of the Group IA metal. A pharmaceutical composition includes a dry formulation of a proton pump inhibitor in a pharmaceutically acceptable carrier including a bicarbonate salt of a Group IA metal. A pharmaceutical composition for making a dry formulation of a proton pump inhibitor which includes a proton pump inhibitor and a bicarbonate salt of a Group IA metal in a form for convenient storage, whereby when the composition is in a dry formulation which is suitable for enteral administration.

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**Description**

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**OMEPRAZOLE SOLUTION AND METHOD OF USING SAME**

This application is a continuation-in-part of United States Serial Number 08/680,376 filed on January 4, 1996.

**TECHNICAL FIELD**

5 The present invention relates to a pharmaceutical preparation containing a substituted benzimidazole, more specifically known as proton pump inhibitor(s) (ppi). More particularly, the present invention relates to a substituted benzimidazole solution/suspension suitable for oral administration.

**BACKGROUND OF THE INVENTION**

Omeprazole is a substituted benzimidazole, 5-methoxy-2-[ (4-methoxy-3,5-dimethyl-2-pyridinyl) methyl] sulfinyl]-1H-benzimidazole, that inhibits gastric acid secretion. Omeprazole belongs to a class of antisecretory compounds, the proton pump inhibitor that do not exhibit anti-cholinergic or H<sub>2</sub> histamine antagonist properties. Drugs of this class suppress gastric acid secretion by the specific inhibition of the H<sup>+</sup>/K<sup>+</sup> ATPase enzyme system (proton pump) at the secretory surface of the gastric parietal cell.

Typically, omeprazole and lansoprazole or other proton pump inhibitors in the form of a delayed-release capsule, is prescribed for short-term treatment of active duodenal ulcers, gastric ulcers, gastroesophageal reflux disease (GERD), severe erosive esophagitis, poorly responsive systematic GERD, and pathological hypersecretory conditions such as Zollinger Ellison syndrome. These conditions are

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caused by an imbalance between acid and pepsin production, called aggressive factors, and mucous, bicarbonate, and prostaglandin production, called defensive factors.

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These above-listed conditions commonly arise in healthy or critically ill patients and may be accompanied by significant upper gastrointestinal bleeding. H<sub>2</sub> antagonists, antacids, and sucralfate are commonly administered to minimize the pain and the complications related to these conditions. These drugs have certain disadvantages associated with their use. Some of these drugs are not completely effective in the treatment of the aforementioned conditions and/or produce adverse side effects, such as mental confusion, constipation, diarrhea, thrombocytopenia, (lowered platelet count) and/or are relatively costly modes of therapy as they require the use of automated infusion pumps for continuous intravenous delivery.

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Patients with significant physiologic stress are at risk for stress-related gastric mucosal damage and subsequent upper gastrointestinal bleeding (Marrone and Silen, 1984). Risk factors that have been clearly associated with the development of stress-related mucosal damage are mechanical ventilation, coagulopathy, extensive burns, head injury, and organ transplant (Zinner et al., 1981; Larson et al., 1984; Czaja et al., 1974; Skillman et al., 1969; and Cook et al., 1994). One or more of these factors are often found in critically ill, intensive care unit patients. A recent cohort study challenges other risk

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factors previously identified such as acid-base disorders, multiple trauma, significant hypertension, major surgery, multiple operative procedures, acute renal failure, sepsis, and coma (Cook et al., 1994). Regardless of the risk type, stress-related mucosal damage results in significant morbidity and mortality. Clinically significant bleeding occurs in at least twenty percent of patients with one or more risk factors who are left untreated (Martin et al., 1993). Of those who bleed, approximately ten percent require surgery (usually gastrectomy) with a reported mortality of thirty percent to fifty percent (Czaja et al., 1974; Peura and Johnson, 1985). Those who do not need surgery often require multiple transfusions and prolonged hospitalization. Prevention of stress-related upper gastrointestinal bleeding is an important clinical goal.

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In addition to general supportive care, the use of drugs to prevent stress-related mucosal damage is considered by many to be the standard of care (AMA Drug Evaluations). However, general consensus is lacking about which drugs to use in this setting (Martin et al., 1993; Gafter et al., 1989; Martin et al., 1992). In two recent meta-analyses (Cook et al., 1991; Tryba, 1994), antacids, sucralfate, and H<sub>2</sub>-antagonists were all found to be superior to placebo and similar to one another in preventing upper gastrointestinal bleeding. Yet, prophylactic agents are withdrawn in fifteen to twenty percent of patients in which they are employed because of failure to prevent bleeding, or control pH (Ostro et al.,

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1985; Siepler, 1986; Ballesteros et al., 1990), or because of adverse effects (Gafer et al., 1989; Sax, 1987; Vial et al., 1991; Cantu and Korek, 1991; Spychal and Wickham, 1985). In addition, the characteristics of an ideal agent for the prophylaxis of stress gastritis and concluded that none of the agents currently in use fulfill their criteria (Smythe and Zarowitz, 1994).

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Omeprazole and lansoprazole and the other proton pump inhibitors reduce gastric acid production by irreversibly inhibiting the H<sup>+</sup>/K<sup>+</sup> ATPase of the parietal cell - the final common pathway for gastric acid secretion (Fellenius et al., 1981; Wallmark et al., 1985; Frylund et al., 1988). Because this drug maintains gastric pH control throughout the dosing interval and has a very good safety profile, it is a logical choice for stress ulcer prophylaxis. The absence of an intravenous or oral liquid dosage form in the United States, however, has limited the testing and use of omeprazole in the critical care patient population. Subsequently, Barie et al. (Barie and Hariri, 1992) described the use of omeprazole enteric-coated pellets administered through a nasogastric tube to control gastrointestinal hemorrhage in a critical care patient with multi-organ failure.

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Stress ulcer prophylaxis has become routine therapy in intensive care units in most hospitals (Fabian et al, 1993.; Cook et al., 1991). Controversy remains regarding pharmacologic intervention to prevent stress-related bleeding in critical care patients. It has been suggested

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that the incidence and risk of gastrointestinal bleeding has decreased in the last ten years and drug therapy may no longer be needed (Cook et al., 1994; Tryba, 1994; Schepp, 1993). This reasoning is not supported by a recent placebo-controlled study. Martin et al. conducted a prospective, randomized, double-blind, placebo-controlled comparison of continuous-infusion cimetidine and placebo for the prophylaxis of stress-related mucosal damage (Marten et al., 1993). The study was terminated early because of excessive bleeding-related mortality in the placebo group. It appears that the natural course of stress-related mucosal damage in a patient at risk who receives no prophylaxis remains significant. In the placebo group, thirty-three percent of patients developed clinically significant bleeding, nine percent required transfusion, and six percent died due to bleeding-related complications. In comparison, fourteen percent of cimetidine-treated patients developed clinically significant bleeding, six percent required transfusions, and 1.5% died due to bleeding-related complication; the difference in bleeding rates between treatment groups was statistically significant. This study clearly demonstrated that continuous-infusion cimetidine reduced morbidity in critical care patients. Although, these data were used to support the approval of continuous-infusion cimetidine by the Food and Drug Administration for stress ulcer prophylaxis, H<sub>2</sub>-antagonists fall short of being the optimal pharmacotherapeutic agents for preventing of stress-related mucosal bleeding.

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Another controversy surrounding stress ulcer prophylaxis is which drug to use. In addition to the various H<sub>2</sub>-antagonists, antacids and sucralfate are other treatment options for the prophylaxis of stress-related mucosal damage. An ideal drug in this setting should possess the following characteristics: prevent stress ulcers and their complications, be devoid of toxicity, lack drug interactions, be selective, have minimal associated costs (such as personnel time and materials), and be easy to administer (Smythe and Zarowitz, 1994).

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Some have suggested that sucralfate is possibly the ideal agent for stress ulcer prophylaxis (Smythe and Zarowitz, 1994). Randomized, controlled studies support the use of sucralfate (Borrero et al., 1986; Tryba, 1987; Cioffi et al., 1994; Driks et al., 1987), but data on critical care patients with head injury, trauma, or burns are limited. In addition, a recent study comparing sucralfate and cimetidine plus antacids for stress ulcer prophylaxis reported clinically significant bleeding in three of forty-eight (6%) sucralfate-treated patients, one of whom required a gastrectomy (Cioffi et al., 1994). In the study performed by Driks and coworkers that compared sucralfate to conventional therapy (H<sub>2</sub>-antagonists, antacids, or H<sub>2</sub>-antagonists plus antacids), the only patient whose death was attributed to stress-related upper gastrointestinal bleeding was in the sucralfate arm (Driks et al., 1987).

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H<sub>2</sub>-antagonists fulfill many of the criteria for an ideal stress ulcer prophylaxis drug. Yet, clinically significant bleeds can occur during H<sub>2</sub>-antagonist prophylaxis (Martin et al., 1993; Cook et al., 1991; Schuman et al., 1987) and adverse events are not uncommon in the critical care population (Gafer et al., 1989; Sax, 1987, Vial et al., 1991; Cantu and Korek, 1991; Spychal and Wickham, 1985). One reason proposed for the therapeutic H<sub>2</sub>-antagonist failures is lack of pH control throughout the treatment period (Ostro et al., 1985). Although the precise pathophysiologic mechanism(s) involved in stress ulceration are not clearly established, the high concentration of hydrogen ions in the mucosa (Fiddian-Green et al., 1987) or gastric fluid in contact with mucosal cells appears to be an important factor. A gastric pH > 3.5 has been associated with a lower incidence of stress-related mucosal damage and bleeding (Larson et al., 1984; Skillman et al., 1969; Skillman et al., 1970; Priebe and Skillman, 1981). Several studies have shown that H<sub>2</sub>-antagonists, even in maximal doses, do not reliably or continuously increase intragastric pH above commonly targeted levels (3.5 to 4.5). This is true especially when used in fixed-dose bolus regimens (Ostro, 1985; Siepler, 1986; Ballesteros et al., 1990). In addition, gastric pH levels tend to trend downward with time when using a continuous-infusion of H<sub>2</sub>-antagonists, which may be the result of tachyphylaxis (Ostro et al., 1985; Wilder-Smith and Merki, 1992).

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Because stress ulcer prophylaxis is frequently employed in the intensive care unit, it is essential from both a clinical and economic standpoint to optimize the pharmacotherapeutic approach. In an attempt to identify optimal therapy, cost of care becomes an issue. All treatment costs should be considered, including the costs of treatment failures and drug-related adverse events. While the actual number of failures resulting in mortality is low, morbidity (e.g., bleeding that requires blood transfusion) can be high, even though its association with the failure of a specific drug is often unrecognized.

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Proton pump inhibitors such as omeprazole represent an advantageous alternative to the use of  $H_2$  antagonists, antacids, and sucralfate as a treatment for complications related to stress-related mucosal damage. However, in their current form (capsules containing an enteric-coated granule formulation of proton pump inhibitor), proton pump inhibitors can be difficult or impossible to administer to patients who are unable (critically ill patients, children, elderly, patients suffering from dysphagia) or patients who are either unwilling or unable to swallow tablets or capsules. Therefore, it would be desirable to formulate a proton pump inhibitor solution or suspension which can be enterally delivered to a patient thereby providing the benefits of the proton pump inhibitor without the drawbacks of the current capsule dose form.

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Omeprazole, the first proton pump inhibitor introduced into use, has been formulated in many different embodiments

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such as in a mixture of polyethylene glycols formed a mixture of adeps solidus and sodium lauryl sulfate in a soluble, basic amino acid to yield a formulation designed for administration in the rectum as shown in United States Patent No. 5,219,870 to Kim.

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United States Patent No. 5,395,323 to Berglund ('323) discloses a device for mixing a pharmaceutical from a solid supply into a parenterally acceptable liquid form for parenteral administration to a patient. The '323 patent teaches the use of an omeprazole tablet which is placed in the device and dissolved by normal saline, and infused into the patient. This device and method of infusing omeprazole does not provide the omeprazole solution as an enteral product nor is this omeprazole solution directly administered to the diseased or affected areas, namely the stomach and upper gastrointestinal tract, nor does this omeprazole formulation provide the immediate anti-acid effect of the present formulation.

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United States Patent No. 4,786,505 to Lovgren et al., discloses a pharmaceutical preparation containing omeprazole together with an alkaline reacting compound or an alkaline salt of omeprazole optionally together with an alkaline compound as a core material in a tablet formulation. The use of the alkaline material, which can be chosen from such substances as the sodium salt of carbonic acid, are used to form a "micro-pH" around each omeprazole particle to protect the omeprazole which is highly sensitive to acid pH. The powder mixture is then

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formulated to small beads, pellets, tablets and may be loaded into capsules by conventional pharmaceutical procedures.

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This formulation of omeprazole does not provide an omeprazole dose form which can be enterally administered to a patient who may be unable and/or unwilling to swallow capsules or pellets nor does it teach a convenient form which can be used to make an omeprazole or other proton pump inhibitor solution or suspension.

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Several buffered omeprazole solutions have been disclosed. Andersson et al., 1993; Landahl et al., 1992; Andersson et al., 1990; Regardh et al., 1990; Andersson et al., 1990; Pilbrant et al., 1985.

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All of the buffered omeprazole solutions described in these references were administered orally and were given to healthy subjects who were able to ingest the oral dose. In all of these studies, omeprazole was suspended in a solution including sodium bicarbonate, as a pH buffer, in order to protect the acid sensitive omeprazole during administration.

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In all of these studies, repeated administration of sodium bicarbonate both prior to, during, and following omeprazole administration were required in order to prevent acid degradation of the omeprazole given via the oral route of administration. As a result, the ingestion of the large amounts of sodium bicarbonate and large volumes of water were required. In the above-cited studies, as much as 48

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mmoles of sodium bicarbonate in 300 ml of water must be ingested for a single dose of omeprazole to be orally administered.

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Initial reports of increased frequency of pneumonia in patients receiving stress ulcer prophylaxis with agents that raise gastric pH has influenced the pharmacotherapeutic approach to management of critical care patients. However, several recent studies (Simms et al., 1991; Pickworth et al., 1993; Ryan et al., 1993; Fabian et al., 1993), a meta-analysis (Cook et al., 1991), and a closer examination of the studies that initiated the elevated pH-associated pneumonia hypotheses (Schepp, 1993) cast doubt on a causal relationship. The relationship between pneumonia and antacid therapy is much stronger than for H<sub>2</sub>-antagonists. The shared effect of antacids and H<sub>2</sub>-antagonists on gastric pH seems an irresistible common cause explanation for nosocomial pneumonia observed during stress ulcer prophylaxis. However, there are important differences between these agents that are not often emphasized (Laggner et al., 1989). When antacids are exclusively used to control pH in the prophylaxis of stress-related upper gastrointestinal bleeding, large volumes are needed. Volume, with or without subsequent reflux, may be the underlying mechanism(s) promoting the development of pneumonia in susceptible patient populations rather than the increased gastric pH. The rate of pneumonia in our study (12%) was not unexpected in this critical care population and compares with sucralfate,

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which does not significantly raise gastric pH (Pickworth et al., 1993; Ryan et al., 1993).

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The buffered omeprazole solutions of the above cited prior art require large amounts of sodium bicarbonate to be given by repeated administration. This is necessary to prevent acid degradation of the omeprazole. The administration of large amounts of sodium bicarbonate can produce at least four significant adverse effects which can dramatically reduce the efficacy of the omeprazole in patients and reduce the overall health of the patients. In the above-cited studies, basically healthy volunteers rather than sick patients were given only one or two dosages of omeprazole utilizing pre-dosing and post-dosing with large volumes of sodium bicarbonate. This dosing protocol would not be suitable for sick or critically ill patients who must receive multiple doses of omeprazole.

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Since bicarbonate is usually neutralized in the stomach or is absorbed, such that belching results, patients with gastroesophageal reflux may exacerbate or worsen their gastroesophageal reflux disease as the belching can cause upward movement of stomach acid (Brunton, 1990).

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Patients with conditions, such as hypertension or heart failure, are standardly advised to avoid the intake of excessive sodium as this can cause aggravation or exacerbation of their hypertensive conditions (Brunton, 1990).

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Additionally, patients with numerous conditions which typically accompany critical illness should avoid the intake of excessive sodium bicarbonate as it can cause metabolic alkalosis which can result in a serious worsening of the patient's condition. Furthermore, excessive antacid intake (such as sodium bicarbonate) can result in drug interactions which produce serious adverse effects. For example, by altering gastric and urinary pH, antacids can alter rates of drug dissolution and absorption, bioavailability, and renal elimination (Brunton, 1990).

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Since buffered omeprazole solution requires prolonged administration of the antacid, sodium bicarbonate, it makes it difficult for patients to comply with the above recommendation.

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15 In addition to the disadvantages associated with excessive intake of sodium bicarbonate, the above-cited prior art teaches a relatively complex regimen for the oral administration of omeprazole. For example, in the Pilbrant et al. (1985) reference, the oral omeprazole administration protocol calls for administering to a subject who has been fasting for at least ten hours, a solution of 8 mmoles of sodium bicarbonate in 50 ml of water. Five minutes later, the subject ingests a suspension of 60 mg of omeprazole in 50 ml of water which also contains 8 mmoles of sodium bicarbonate. This is rinsed down with another 50 ml of 8 mmoles sodium bicarbonate solution. Ten minutes after the ingestion of the omeprazole dose, the subject ingests 50 ml of bicarbonate solution (8 mmoles). This is repeated at

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twenty minutes and thirty minutes post omeprazole dosing to yield a total of 48 mmoles of sodium bicarbonate and 300 ml of water in total which are ingested by the subject for a single omeprazole dose.

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5 Not only does this regimen require the ingestion of excessive amounts of bicarbonate and water, it is unlikely that a healthy patient would comply with this regimen for each dose of omeprazole over the course of a prescribed omeprazole protocol. It is unlikely or even improbable that a critically ill patient would be able to comply with this regimen.

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Even in healthy patients, the complexity of the drug regimen leads to the conclusion that patients would be unlikely to comply with this regimen thereby leading to a lack of beneficial outcome for the patient. It is well documented that patients who are required to follow complex schedules for drug administration are non-compliant and, thus, the efficacy of the buffered omeprazole solutions of the prior art would be expected to be reduced due to non-compliance. Compliance has been found to be markedly reduced when patients are required to deviate from a schedule of one or two (usually morning and night) doses of a medication per day. The use of the prior art buffered omeprazole solutions which require administration protocols with numerous steps, different drugs (sodium bicarbonate + omeprazole + PEG400 versus sodium bicarbonate alone), and specific time allotments between each stage of the total omeprazole regimen in order to achieve efficacious results

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is clearly in contrast with both current drug compliance theories and human nature.

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The prior art (Pilbrant et al., 1985) teaches that the buffered omeprazole suspension can be stored at refrigerator temperatures for a week and deep frozen for a year while still maintaining 99% of their initial potency. It would be desirable to have an omeprazole or other proton pump inhibitor solution or suspension which could be stored at room temperature or in a refrigerator for periods of time which exceed those of the prior art while still maintaining 99% of the initial potency. Additionally, it would be advantageous to have a form of the omeprazole and bicarbonate which can be utilized to instantly make the omeprazole solution/suspension of the present invention which is supplied in a solid form which imparts the advantages of improved shelf-life at room temperature, lower cost to produce, less expensive shipping costs, and which is less expensive to store.

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The present invention provides a solution/suspension of proton pump inhibitors such as omeprazole, lansoprazole or other proton pump inhibitor (pantoprazole, rabeprazole, dantoprazole, nabeproazole, perprazole or other proton pump inhibitor) which is suitable for administration which includes all of the aforementioned advantages.

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It would, therefore, be desirable to have an proton pump inhibitor formulation which provides a cost effective means for the treatment of the aforementioned conditions

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without the adverse effect profile of  $H_2$  receptor antagonist, antacids, and sucralfate. Further, it would be desirable to have a proton pump inhibitor formulation which is convenient to prepare and administer to patients unable to ingest capsules, which is rapidly absorbed, can be orally or enterally delivered as a liquid form or solid form which becomes a liquid in the stomach or upper GI tract (the desired treatment regimen.). It is desirable that the liquid formulation not clog indwelling tubes, such as nasogastric tubes or other similar tubes, and which acts as an antacid immediately upon delivery. Furthermore, it would be desirable to have a pharmaceutical composition which is highly efficacious for the treatment of the aforementioned conditions.

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#### SUMMARY OF THE INVENTION AND ADVANTAGES

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In accordance with the present invention, there is provided a method of treating gastric acid disorders by administering to a patient a pharmaceutical composition including a proton pump inhibitor in a pharmaceutically acceptable carrier including a bicarbonate salt of a Group IA metal where the administering step consists of a single dosage form without requiring further administering of the bicarbonate salt of the Group IA metal.

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The present invention further provides a pharmaceutical composition includes a dry formulation of a proton pump inhibitor in a pharmaceutically acceptable carrier including a bicarbonate salt of a Group IA metal.

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The present invention further provides a pharmaceutical composition for making a dry formulation of a proton pump inhibitor which includes a proton pump inhibitor and a bicarbonate salt of a Group IA metal in a form for convenient storage, whereby the composition is in a dry formulation which is suitable for enteral administration.

#### BRIEF DESCRIPTION OF THE DRAWINGS

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Other advantages of the present invention will be readily appreciated as the same becomes better understood by reference to the following detailed description when considered in connection with the accompanying drawing wherein:

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Figure 1 is a graph showing the effect of the omeprazole solution/suspension of the present invention on gastric pH in patients at risk for upper gastrointestinal bleeding from stress-related mucosal damage;

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Figure 2 is a flow chart illustrating a patient enrollment scheme;

Figure 3 is a bar graph illustrating gastric pH both pre- and post-administration of omeprazole solution/suspension according to the present invention; and

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Figure 4 is a graph illustrating the stomach pH values of both chocolate plus lansoprazole in powder form and lansoprazole alone.

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## DETAILED DESCRIPTION OF THE INVENTION

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A pharmaceutical composition which can include an aqueous solution/suspension, or dry formulation, of proton pump inhibitors, such as omeprazole or other substituted benzimidazoles which are proton pump inhibitors such as lansoprazole, pantoprazole, rabeprazole, dontoprazole, perprazole, habeprazole, and derivatives thereof in a pharmaceutically acceptable carrier including a bicarbonate salt of a Group IA metal is disclosed. For the purposes of description, the composition includes dry formulations, solutions and/or suspensions of the omeprazole or other proton pump inhibitors. Hereinafter, the use of the term "solution" includes solutions and/or suspensions of the substituted benzimidazoles.

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10 description, the composition includes dry formulations, solutions and/or suspensions of the omeprazole or other proton pump inhibitors. Hereinafter, the use of the term "solution" includes solutions and/or suspensions of the substituted benzimidazoles.

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15 The pharmaceutical composition of the present invention is prepared by mixing omeprazole (Merck & Co. Inc., West Point, PA) or other proton pump inhibitors or derivatives thereof with a solution including a bicarbonate salt of a Group IA metal. Preferably, omeprazole or other proton pump inhibitor powder or granules, which can be obtained from a capsule, are mixed with a sodium bicarbonate solution to achieve a desired final omeprazole concentration. As an example, the concentration of omeprazole in the solution/suspension can range from approximately 0.5 mg/ml to approximately 6.0 mg/ml. The preferred concentration for the omeprazole in the solution/suspension ranges from approximately 1.0 mg/ml to

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20 proton pump inhibitor powder or granules, which can be obtained from a capsule, are mixed with a sodium bicarbonate solution to achieve a desired final omeprazole concentration. As an example, the concentration of omeprazole in the solution/suspension can range from approximately 0.5 mg/ml to approximately 6.0 mg/ml. The preferred concentration for the omeprazole in the solution/suspension ranges from approximately 1.0 mg/ml to

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approximately 4.0 mg/ml with 2mg/ml being the standard concentration.

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The pharmaceutically effective carrier includes the bicarbonate salt of the Group IA metal and can be prepared by mixing the bicarbonate salt of the Group IA metal, preferably sodium bicarbonate, with water. The concentration of the bicarbonate salt of the Group IA metal in the composition generally ranges from approximately 5.0 percent to approximately 60.0 percent. Preferably, the concentration of the bicarbonate salt of the Group IA metal ranges from approximately 7.5 percent to approximately 10.0 percent. In a preferred embodiment of the present invention, sodium bicarbonate is the preferred salt of the Group IA metal and is present in a concentration of approximately 8.4 percent.

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In a preferred embodiment of the present invention, specific example of the enterically-coated omeprazole particles are obtained from delayed release capsules (Astra Merck) alternatively omeprazole powder can be used. The coated omeprazole particles are mixed with a sodium bicarbonate ( $\text{NaHCO}_3$ ) solution which dissolves the enteric coating and forms an omeprazole solution/suspension in accordance with the present invention. The omeprazole solution/suspension has significant pharmacokinetic advantages over standard time-release omeprazole capsules including: a decreased drug absorbance time (-10 to 12 minutes) following administration for the omeprazole solution versus (-2-3 hours) following administration for

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the enteric coated pellets; the  $\text{NaHCO}_3$  solution protects the omeprazole from acid degradation prior to absorption; the  $\text{NaHCO}_3$  acts as an antacid while the omeprazole is being absorbed; and the solution/suspension can be administered through an existing indwelling tube without clogging, for example, nasogastric or other feeding tubes (jejunal or duodenal) including small bore needle catheter feeding tubes.

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As stated above, other proton pump inhibitors can be substituted for the omeprazole or lansoprazole, or other suitable substituted benzimidazoles without departing from the spirit of the present invention. These proton pump inhibitors can include, but are not limited to, lansoprazole, pantoprazole, rabeprazole, omeprazole, perprazole, habeprazole and other proton pump or acid pump inhibitors. Proton pump inhibitors are membrane impermeable and form disulfide covalent bonds with cysteine residues in the alpha subunit which inhibit the activity of the acid secreting pump. The data presented provides a factual basis for suggesting the use of this group of proton pump inhibitor compounds in accordance with the present invention.

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The pharmaceutical composition including the proton pump inhibitors such as omeprazole or lansoprazole or other proton pump inhibitors and derivatives thereof in a pharmaceutically acceptable carrier of a bicarbonate salt of Group IA metal can be used for the treatment of gastrointestinal conditions including, but not limited to,

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active duodenal ulcers, gastric ulcers, gastroesophageal reflux disease (GERD), severe erosive esophagitis, poorly responsive systematic GERD, and pathological hypersecretory conditions such as Zollinger Ellison Syndrome. These conditions are caused by imbalances between acid and pepsin production, called aggressive factors, and mucous, bicarbonate, and prostaglandin production, called defensive factors. Treatment of these conditions is accomplished by administering to a patient an effective amount of the pharmaceutical composition according to the present invention.

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The proton pump inhibitor is administered and dosed in accordance with good medical practice, taking into account the clinical condition of the individual patient, the site and method of administration, scheduling of administration, and other factors known to medical practitioners. The "effective amount" for purposes herein thus determine by such considerations as are known in the art. The amount must be effective to achieve improvement, including but not limited to, raising of gastric pH, reduced gastrointestinal bleeding, reduction in the need for blood transfusion, improved survival rate, more rapid recovery, or improvement or elimination of systems and other indicators as are selected as appropriate measures by those skilled in the art.

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The dosage range of omeprazole or other proton pump inhibitors such as substituted benzimidazoles and derivatives thereof can range from approximately 2 mg/day

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to approximately 100 mg/day. The standard daily dosage is typically 20 mg omeprazole in 10 ml of solution.

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In the method of the present invention, the omeprazole solution suspension can be administered in various ways.

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5 It should be noted that the omeprazole solution/suspension can be administered as the compound or as the pharmaceutically acceptable salt form (dry) and can be administered alone or in combination with pharmaceutically acceptable carriers. The compounds can be administered orally or enterally. The formulations can be made more palatable by adding flavorings such as chocolate, root beer, and others.

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Additionally, various additives including ambicin which enhance the stability, sterility, and isotonicity of the compositions. Additionally, antimicrobial preservatives, antioxidants, chelating agents, and buffers can be added. However, microbiological evidence shows that this formulation inherently possesses anti-microbial activity. Prevention of the action of microorganisms can be enhanced by various antibacterial and antifungal agents, for example, parabens, chlorobutanol, phenol, sorbic acid, and the like.

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In many cases, it would be desirable to include isotonic agents, for example, sugars, sodium chloride, and the like. Additionally, thickening agents, such as methyl cellulose, are desirable to use in order to reduce settling the omeprazole or derivatives thereof from the suspension.

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Furthermore, the use of flavoring agents such as chocolate or chelating agents such as calcium have been shown to act as a potentiator or enhancer of the pharmacologic activity of the proton pump inhibitor as is shown in Figure 4. More specifically, Figure 4 shows that the gastric pH, after 18 hours, of the lansoprazole alone was approximately 1.5, whereas the lansoprazole with the chocolate was approximately 2.9. Therefore the combination of the lansoprazole with chocolate enhanced the pharmacologic activity of the lansoprazole. By measuring the serum gastrin the results establish that the sodium bicarbonate as well as chocolate flavoring and calcium were all able to stimulate the activation of the proton pumps, perhaps due to the release of gastrin. Proton pump inhibitors work by functionally inhibiting the proton pump and can only block activated proton pumps. By further administering the proton pump inhibitor with one of these potentiators or enhancers, there is a synchronization of activation of the proton pump with the absorption of the proton pump inhibitor. This combination produced a much longer pharmacologic effect than when the proton pump inhibitor was administered alone.

The formulations of the present invention can also be manufactured in a concentrated form, such as an effervescent tablet or powder, so that upon reaction with water, the aqueous form of the present invention would be produced for oral or enteral administration.

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Additionally, the present invention can be manufactured by utilizing micronized compounds in place of the granules or powder. This process is known as micronization and is utilized in order to produce a particle having a greater diameter. Micronization is the process by which solid drug particles are reduced in size. Since the dissolution rate is directly proportional to the surface area of the solid, and reducing the particle size increases the surface area, reducing the particle size increases the dissolution rate.

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Although micronization results in increased surface area causing particle aggregation, which can negate the benefit of micronization and is an expensive manufacturing step, it does have the significant benefit of increasing the dissolution rate of relatively water insoluble drugs, such as omeprazole and other proton pump inhibitors.

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A pharmacological formulation of the proton pump inhibitors utilized in the present invention can be administered orally to the patient. A pharmacological formulation of the omeprazole solution/suspension utilized in the present invention is preferably administered enterally. This can be accomplished, for example, by administering the solution/suspension via a nasogastric tube or other indwelling tubes. In order to avoid the critical disadvantages associated with administering large amounts of sodium bicarbonate, the omeprazole solution of the present invention is administered in a single dose which does not require any further administration of

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bicarbonate following the administration of the omeprazole solution. That is, unlike the prior art omeprazole solutions and administration protocols outlined above, the formulation of the present invention is given in a single dose which does not require administration of bicarbonate either before administration of the omeprazole or after administration of the omeprazole. The present invention eliminates the need to pre-or post-dose with additional volumes of water and sodium bicarbonate. The amount of bicarbonate administered via the single dose administration of the present invention is less than the amount of bicarbonate administered as taught in the prior art references cited above.

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The amount of sodium bicarbonate used in the solution/suspension of the present invention is approximately 1 meq (or mmole) sodium bicarbonate per 2 mg omeprazole, with a range of approximately 0.75 meq (mmole) to 1.5 meq (mmole) per 2 mg of omeprazole.

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The present invention further includes a pharmaceutical composition for making a solution/suspension of proton pump inhibitors, which consists essentially of omeprazole or other proton pump inhibitors and derivatives thereof and a bicarbonate salt of a Group IA metal in a form convenient for storage, whereby when the composition is placed into an aqueous solution, the composition dissolves yielding a solution/suspension suitable for enteral administration to a subject. The pharmaceutical composition is in a solid form prior to dissolution in the

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aqueous solution. The omeprazole or other substituted benzimidazoles and derivatives thereof and bicarbonate can be formed into a tablet, capsules, or granules, by methods well known to those skilled in the art.

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5 The pharmaceutical composition suitable for making a solution/suspension according to the present invention can further include an effervescing agent to aid in the dissolution of the pharmaceutical composition in the aqueous solution. In the present invention the effervescing agent is sodium bicarbonate.

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The resultant omeprazole solution is stable at room temperature for several weeks and inhibits the growth of bacteria or fungi as shown in Example IV below. By providing a pharmaceutical composition including the omeprazole or other substituted benzimidazole and derivatives thereof with bicarbonate in a solid form, which is dissolved in a prescribed amount of aqueous solution to yield the desired concentration of omeprazole and bicarbonate, the cost of production, shipping, and storage are greatly reduced as no liquids are shipped (reducing weight and cost) and there is no need to refrigerate the solid form of the composition or the solution. The resultant solution, can be formulated and then used to provide dosages for a single patient over a course of time or for several patients.

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The solid formulation of the present invention could be in the form of a powder, a tablet, a capsule, or other

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suitable solid dosage form (e.g. a pelleted form or an effervescent tablet or powder). The solid formulation would then create the present invention when acted upon by a suitable vehicle, for example water. The water may be added either prior to ingestion or the dry formulation may be ingested first and then acted upon by the water utilized to swallow the solid formulation. A third mechanism enables water in the stomach secretions to produce the present invention.

10 The following experimental data illustrate the utility of the pharmaceutical composition of the present invention.

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#### METHODS

##### Example I

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15 Patients were evaluable if they met the following criteria: had two or more risk factors for SRMD (mechanical ventilation, head injury, severe burn, sepsis, multiple trauma, adult respiratory distress syndrome, major surgery, acute renal failure, multiple operative procedures, coagulotherapy, significant hypotension, acid-base disorder, and hepatic failure), gastric pH of  $\leq 4$  prior to study entry, and no concomitant prophylaxis for SRMD.

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25 Nasogastric (ng) tubes were placed in the patients and an omeprazole dosage protocol of 40 mg omeprazole solution/suspension followed by 40 mg omeprazole solution/suspension in eight hours, then 20 mg omeprazole solution/suspension per day, for five days. After each

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omeprazole solution/suspension administration, nasogastric suction was turned off for thirty minutes.

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Results:

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Eleven patients were evaluable. All patients were mechanically ventilated. Two hours after the initial dose of omeprazole solution/suspension 40 mg omeprazole, all patients had an increase in gastric pH to greater than eight as shown in Figure 1. Ten of the eleven patients maintained a gastric pH of greater than or equal to four on 20 mg omeprazole solution/suspension. One patient required 40 mg omeprazole solution/suspension per day (closed head injury, five total risk factors for SRMD). Two patients were changed to omeprazole solution/suspension after having developed clinically significant upper gastrointestinal bleeding while receiving conventional intravenous H<sub>2</sub> antagonists. Bleeding subsided in both cases after twenty-four hours. Clinically significant upper gastrointestinal bleeding did not occur in the other nine patients. Overall mortality was 27%, mortality attributable to upper gastrointestinal bleeding was 0%. Pneumonia developed in one patient after initiating omeprazole therapy and was present upon the initiation of omeprazole therapy in another patient. The mean length of prophylaxis was five days.

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A pharmacoeconomic analysis revealed a difference in the total cost of care for the prophylaxis of SRMD:

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ranitidine (Zantac®) continuous infusion intravenously  
(150 mg/24 hours) x five days \$125.50;

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cimetidine (Tagamet®) continuous infusion  
intravenously (900 mg/24 hours) x five days \$109.61;

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5 sucralfate one gm slurry four times a day per (ng)  
tube x five days \$73.00; and

SOS regimen per (ng) tube x five days \$65.70.

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#### Conclusion:

This example illustrates the efficacy of the  
simplified omeprazole solution of the present invention  
based on the increase in gastric pH, safety and  
cost/convenience of the omeprazole solution/suspension as a  
method for SRMD prophylaxis.

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#### Example II

15 Experiments were carried out in order to determine the  
effect of the omeprazole solution/suspension  
(omeprazole/sodium bicarbonate solution) administration on  
the accuracy on subsequent pH measurements through a  
nasogastric tube.

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#### 20 Methods:

The omeprazole solution/suspension was prepared by  
mixing 10 ml of 8.4% sodium bicarbonate with the contents  
of a 20 mg capsule of omeprazole (Merck & Co. Inc., West  
Point, PA) to yield a solution/suspension having a final  
omeprazole concentration of 2 mg/ml. After mixing the

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omeprazole solution/suspension, it was administered into the stomach, usually, through a nasogastric (ng) tube. Nasogastric tubes from nine different institutions were gathered for an evaluation 400 mg omeprazole solution/suspension was prepared as described above. Artificial gastric fluid (gf) was prepared according to the USP. pH recordings were made in triplicate using a Microcomputer Portable pH meter model 6007 (Jenco Electronics Ltd., Taipei, Taiwan). [1] First the terminal portion (tp) of the nasogastric tubes was placed into a glass beaker containing the gastric fluid. A 5 ml aliquot of gastric fluid was aspirated through each tube and the pH recorded, this was called the "pre-omeprazole solution/suspension measurement". [2] Secondly, the terminal portion (tp) of each of the nasogastric tubes was removed from the beaker of gastric fluid and placed into an empty beaker. Twenty (20) mg of omeprazole solution/suspension was delivered through each of the nasogastric tubes and flushed with 10 ml of tap water. The terminal portion (tp) of each of the nasogastric tubes was placed back into the gastric fluid. After a one hour incubation, a 5 ml aliquot of gastric fluid was aspirated through each nasogastric tube and the pH recorded, this was called the "after 1st dose SOS measurement". [3] After an additional hour had passed, the second step was repeated, this was called the "after 2nd ND dose SOS measurement". In addition to the pre-SOS measurement, the pH of the gastric fluid was checked in triplicate after steps [2] and [3]. A change in the pH measurements of +/- 0.3 units was

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considered significant. The Friedman test was used to compare the results. The Friedman test is a two way analysis of variance which is used when more than two related samples are of interest, as in repeated measurements.

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#### Results:

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The results of this experiments are outlined in Table 1. Table 1 illustrates the results of the pH measurements that were taken during the course of the experiment. These results illustrate that there were no statistically significantly latent effects of omeprazole solution/suspension administration (per nasogastric tube) on the accuracy of subsequent pH measurements obtained through the same nasogastric tube.

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#### Example III

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Experiments were performed in order to determine the efficacy, safety, and cost of simplified omeprazole suspension in mechanically ventilated critically ill patients who have at least one additional risk factor for stress-related mucosal damage.

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#### Methods:

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**Patients:** Seventy-five adult, mechanically ventilated patients with at least one additional risk factor for stress-related mucosal damage.

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**Interventions:** Patients received 20 ml omeprazole suspension (containing 40 mg of omeprazole) initially,

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followed by a second 20 ml dose six-eight hours later, then 10 ml (20 mg) daily. Omeprazole solution/suspension according to the present invention was administered through a nasogastric tube, followed by 5-10 ml of tap water. The nasogastric tube was clamped for one-two hours after each administration.

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Measurements and Main Results: The primary outcome measure was clinically significant gastrointestinal bleeding determined by endoscopic evaluation, nasogastric aspirate examination, or heme-positive coffee ground material that did not clear with lavage and was associated with a five percent decrease in hematocrit. Secondary efficacy measures were gastric pH measured four hours after omeprazole was first administered, mean gastric pH after omeprazole was started, and the lowest gastric pH during omeprazole therapy. Safety-related outcomes included the incidence of adverse events and the incidence of pneumonia. No patient experienced clinically significant upper gastrointestinal bleeding after receiving omeprazole suspension. The four-hour post omeprazole gastric pH was 7.1 (mean), the mean gastric pH after starting omeprazole was 6.8 (mean) and the lowest pH after starting omeprazole was 5.6 (mean). The incidence of pneumonia was twelve percent. No patient in this high-risk population experienced an adverse event or a drug interaction that was attributable to omeprazole.

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Conclusions: Omeprazole suspension prevented clinically significant upper gastrointestinal bleeding and

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maintained gastric pH above 5.5 in mechanically ventilated critical care patients without producing toxicity.

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Materials and Methods:

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The study protocol was approved by the Institutional Review Board for the University of Missouri at Columbia.

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*Study Population:* All adult (>18 years old) patients admitted to the surgical intensive care and burn unit at the University of Missouri Hospital with an intact stomach, a nasogastric tube in place, and an anticipated intensive care unit stay of at least forty-eight hours were considered for inclusion in the study. To be included patients also had to have a gastric pH of <4, had to be mechanically ventilated and have one of the following additional risk factors for a minimum of twenty-four hours after initiation of omeprazole suspension: head injury with altered level of consciousness, extensive burns (>20% Body Surface Area), acute renal failure, acid-base disorder, multiple trauma, coagulopathy, multiple operative procedures, coma, hypotension for longer than one hour or sepsis (see Table 2). Sepsis was defined as the presence of invasive pathogenic organisms or their toxins in blood or tissues resulting in a systematic response that included two or more of the following: temperature greater than 38°C or less than 36°C, heart rate greater than 90 beats/minute, respiratory rate greater than 20 breaths/minute (or  $P_{O_2}$  less than 75 mm Hg), and white blood cell count greater than 12,000 or less than 4000 cells/mm<sup>3</sup> or more than 10 percent bands (Bone, 1991). Patients in whom H<sub>2</sub>-antagonist

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therapy had failed or who experienced an adverse event while receiving H<sub>2</sub>-antagonist therapy were also included.

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Patients were excluded from the study if they were receiving azole antifungal agents through the nasogastric tube; were likely to swallow blood (e.g., facial and/or sinus fractures, oral lacerations); had severe thrombocytopenia (platelet count less than 30,000 cells/mm<sup>3</sup>); were receiving enteral feedings through the nasogastric tube; or had a history of vagotomy, pyloroplasty, or gastroplasty. In addition, patients with a gastric pH above four for forty-eight hours after ICU admission (without prophylaxis) were not eligible for participation. Patients who developed bleeding within the digestive tract that was not stress-related mucosal damage (e.g., endoscopically verified variceal bleeding or Mallory-Weiss tears, oral lesions, nasal tears due to placement of the nasogastric tube) were excluded from the efficacy evaluation and categorized as having non-stress-related mucosal bleeding. The reason for this exclusion is the confounding effect of non-stress-related mucosal bleeding on efficacy-related outcomes, such as the use of nasogastric aspirate inspection to define clinically significant upper gastrointestinal bleeding.

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Study Drug Administration: Omeprazole solution/suspension was prepared immediately before administration by the patient's nurse using the following instructions: 1) Empty the contents of one or two 20 mg omeprazole capsule(s) into an empty 10 ml syringe (with 20

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gauge needle in place) from which the plunger has been removed. (Omeprazole delayed-release capsules, Merck & Co., Inc., West Point, PA). 2) Replace the plunger and uncap the needle. 3) Withdraw 10 ml of 8.4% sodium bicarbonate solution or 20 ml if 40 mg given (Abbott Laboratories, North Chicago, IL). The resultant preparation should contain 2 mg omeprazole per ml of 8.4% sodium bicarbonate. 4) Allow the enteric coated pellets of omeprazole to completely breakdown, 30 minutes (agitation is helpful). The omeprazole in the resultant preparation is partially dissolved and partially suspended. The preparation should have a milky white appearance with fine sediment and should be shaken before using. The solution/suspension was not administered with acidic substances. A high pressure liquid chromatography study was performed that has demonstrated that this preparation of simplified omeprazole suspension maintains >90% potency for seven days at room temperature. This preparation remained free of bacterial and fungal contamination for thirty days when stored at room temperature (see Table 5).

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The initial dose of omeprazole solution/suspension was 40 mg, followed by a second 40 mg dose 6-8 hours later, then a 20 mg daily dose administered at 8:00 AM. Each dose was administered through the nasogastric tube. The nasogastric tube was then flushed with 5-10 ml of tap water and clamped for at least one hour. Omeprazole therapy was continued until there was no longer a need for stress ulcer prophylaxis (usually after the nasogastric tube removed and

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the patient was taking water/food by mouth, or after the patient was removed from mechanical ventilation).

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*Primary Outcome Measures:* The primary outcome measure in this study was the rate of clinically significant stress-related mucosal bleeding defined as endoscopic evidence of stress-related mucosal bleeding or bright red blood per nasogastric tube that did not clear after a 5-minute lavage or persistent Gastrocult (SmithKline Diagnostics, Sunnyville, CA) positive coffee ground material for four consecutive hours that did not clear with lavage (at least 100 ml) and produced a 5% decrease in hematocrit.

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*Secondary Outcome Measures:* The secondary efficacy measures were gastric pH measured four hours after omeprazole was administered, mean gastric pH after starting omeprazole and lowest gastric pH during omeprazole administration. Gastric pH was measured immediately after aspirating gastric contents through the nasogastric tube. pH paper (pHydrion improved pH papers, Microessential Laboratory, Brooklyn, NY) was used to measure gastric aspirate pH. The pH range of the test strips was 1 to 11, in increments of one pH unit. Gastric pH was measured before the initiation of omeprazole solution/suspension therapy, immediately before each dose, and every four hours between doses.

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Other secondary outcome measures were incidence of adverse events (including drug interactions) and pneumonia.

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Any adverse event that developed during the study was recorded. Pneumonia was defined using indicators adapted from the Centers for Disease Prevention and Control definition of nosocomial pneumonia (Garner et al., 1988).

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5 According to these criteria, a patient who has pneumonia is one who has rales or dullness to percussion on physical examination of the chest or has a chest radiograph that shows new or progressive infiltrate(s), consolidation, cavitation, or pleural effusion and has at least two of the  
20 following present: new purulent sputum or changes in character of the sputum, an organism isolated from blood culture, fever or leukocytosis, or evidence of infection from a protective specimen brush or bronchoalveolar lavage.  
25 Patients who met the criteria for pneumonia and were receiving antimicrobial agents for the treatment of pneumonia were included in the pneumonia incidence figure. These criteria were also used as an initial screen before the first dose of study drug was administered to determine if pneumonia was present prior to the start of omeprazole  
35 suspension.

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Cost of Care Analysis: A pharmacoeconomic evaluation of stress ulcer prophylaxis using omeprazole solution/suspension was performed. The evaluation included total drug cost (acquisition and administration), actual  
25 costs associated with adverse events (e.g., psychiatry consultation for mental confusion), costs associated with clinically significant upper gastrointestinal bleeding. Total drug cost was calculated by adding the average

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institutional costs of omeprazole 20 mg capsules, 50 ml sodium bicarbonate vials, and 10 ml syringes with needle; nursing time (drug administration, pH monitoring); pharmacy time (drug preparation); and disposal costs. Costs associated with clinically significant upper gastrointestinal bleeding included endoscopy charges and accompanying consultation fees, procedures required to stop the bleeding (e.g., surgery, hemostatic agents, endoscopic procedures), increased hospital length of stay (as assessed by the attending physician), and cost of drugs used to treat the gastrointestinal bleeding.

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Statistical Analysis: The paired t-test (two-tailed) was used to compare gastric pH before and after omeprazole solution/suspension administration and to compare gastric pH before omeprazole solution/suspension administration with the mean and lowest gastric pH value measured after beginning omeprazole.

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#### Results:

Seventy-seven patients met the inclusion and exclusion criteria and received omeprazole solution/suspension (see Figure 2). Two patients were excluded from the efficacy evaluation because the protocol for omeprazole administration was not followed. In one case, the omeprazole enteric-coated pellets had not completely broken down prior to the administration of the first two doses, which produced an erratic effect on gastric pH. The gastric pH increased to above six as soon as the patient was given a dose of omeprazole solution/suspension (in

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which the enteric coated pellets of omeprazole had been allowed to completely breakdown).

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The reason for the second exclusion was that nasogastric suctioning was not turned off after the omeprazole dose was administered. This resulted in a transient effect on gastric pH. The suction was turned off with subsequent omeprazole doses, and control of gastric pH was achieved. Two patients were considered efficacy failures because omeprazole failed to maintain adequate gastric pH control on the standard omeprazole 20 mg/day maintenance dose. When the omeprazole dose was increased to 40 mg/day (40 mg once/day or 20 mg twice/day), gastric pH was maintained above four in both patients. These two patients were included in the safety and efficacy evaluations, including the gastric pH analysis. After the two patients were declared failures, their pH values were no longer followed.

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The ages of the remaining seventy-five patients ranged from eighteen to eighty-seven years; forty-two patients were male and thirty-three were female. All patients were mechanically ventilated during the study. Table 2 shows the frequency of risk factors for stress-related bleeding that were exhibited by the patients in this study. The most common risk factors in this population were mechanical ventilation and major surgery. The range of risk factors for any given patient was two to ten, with a mean of 3 ( $\pm 1$ ) (standard deviation). Five patients enrolled in the study had developed clinically significant bleeding while

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receiving continuous infusions of ranitidine (150 mg/24 hr) or cimetidine (900 mg/24 hr). In all five cases, the bleeding subsided and the gastric pH rose to above five within thirty-six hours after initiating omeprazole therapy. Three patients were enrolled after having developed two consecutive gastric pH values below three while receiving an H<sub>2</sub>-antagonist (in the doses outlined above). In all three cases, gastric pH rose to above five within four hours after omeprazole therapy was initiated. Four other patients were enrolled in this study after experiencing confusion (n=2) or thrombocytopenia (n=2) during H<sub>2</sub>-antigens therapy. Within thirty-six hours of switching therapy, these adverse events resolved.

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*Stress-related Mucosal Bleeding and Mortality:* None of the sixty-five patients who received simplified omeprazole suspension as their initial prophylaxis against stress-related mucosal bleeding developed overt or clinically significant upper gastrointestinal bleeding. In four of the five patients who had developed upper gastrointestinal bleeding before study entry, bleeding diminished to the presence of occult blood only (Gastroccult-positive) within eighteen hours of starting omeprazole suspension; bleeding stopped in all patients within thirty-six hours. The overall mortality rate in this group of critically ill patients was eleven percent. No death was attributable to upper gastrointestinal bleeding or the use of omeprazole solution/suspension.

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*Gastric pH:* The mean ( $\pm$  standard deviation) pre-omeprazole gastric pH was  $3.5 \pm 1.9$ . Within four hours of omeprazole administration, the gastric pH rose to  $7.1 \pm 1.1$  (see Figure 3); this difference was significant ( $p < 0.001$ ). The differences between pre-omeprazole gastric pH and the mean and lowest gastric pH measurements during omeprazole administration ( $6.8 \pm 0.6$  and  $5.6 \pm 1.3$ , respectively) were also statistically significant ( $p < 0.001$ ).

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*Safety:* Omeprazole solution/suspension was well tolerated in this group of critically ill patients. Only one patient with sepsis experienced an adverse event that may have been drug-related thrombocytopenia. However, the platelet count continued to fall after omeprazole was stopped. The platelet count then returned to normal despite reinstitution of omeprazole therapy. Of note, one patient on a jet ventilator continuously expelled all liquids placed in her stomach up and out through her mouth, and thus was unable to continue on omeprazole. No clinically significant drug interactions with omeprazole were noted during the study period. As stated above, metabolic alkalosis is a potential concern in patients receiving sodium bicarbonate. However, the amount of sodium bicarbonate in omeprazole solution/suspension was small ( $12 \text{ mEq}/10 \text{ ml}$ ) and no electrolyte abnormalities were found.

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*Pneumonia:* Pneumonia developed in nine (12%) patients receiving omeprazole solution/suspension. Pneumonia was

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present in an additional five patients before the start of omeprazole therapy.

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Pharmacoeconomic evaluation: The average length of treatment was nine days. The cost of care data are listed in Tables 3 and 4. The costs of drug acquisition, preparation, and delivery for some of the traditional agents used in the prophylaxis of stress-related upper gastrointestinal bleeding are listed in Table 3. There were no costs to add from toxicity associated with omeprazole solution/suspension. Since two of seventy-five patients required 40 mg of omeprazole solution/suspension daily to adequately control gastric pH, the acquisition/preparation cost should reflect this. The additional 20 mg of omeprazole with vehicle adds seven cents per day to the cost of care. Therefore, the daily cost of care for omeprazole solution/suspension in the prophylaxis of stress-related mucosal bleeding was \$12.60 see Table 4.

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Omeprazole solution/suspension is a safe and effective therapy for the prevention of clinically significant stress-related mucosal bleeding in critical care patients. The contribution of many risk factors to stress-related mucosal damage has been challenged recently (6). All of the patients in this study had at least one risk factor that has clearly been associated with stress-related mucosal damage - mechanical ventilation. Previous trials and data from a recently published study show that stress ulcer prophylaxis is of proven benefit in patients at risk

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and, therefore, it was thought to be unethical to include a placebo group in this study. No clinically significant upper gastrointestinal bleeding occurred during omeprazole solution/suspension therapy. Gastric pH was maintained above 4 on omeprazole 20 mg/day in seventy-three of seventy-five patients. No adverse events or drug interaction associated with omeprazole were encountered.

#### Example IV

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The anti-microbial or bacteriostatic effects of the omeprazole solution/suspension were analyzed by applicants.

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An omeprazole solution/suspension made according to the present invention was stored at room temperature for four weeks and then was analyzed for fungal and bacterial growth.

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#### Results:

Following four weeks of storage at room temperature, no bacterial or fungal growth was detected.

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An omeprazole solution/suspension made in accordance with the present invention was stored at room temperature for twelve weeks and then was analyzed for fungal and bacterial growth.

#### Results:

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After twelve weeks of incubation at room temperature, no fungal or bacterial growth was detected.

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The results of these experiments illustrate the stability and bacteriostatic characteristics of the omeprazole solution/suspension of the present invention.

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Throughout this application various publications and patents are referenced by citation and number. Full citations for the publication are listed below. The disclosure of these publications and patents in their entireties are hereby incorporated by reference into this application in order to more fully describe the state of the art to which this invention pertains.

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The invention has been described in an illustrative manner, and it is to be understood the terminology used is intended to be in the nature of description rather than of limitation.

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Obviously, many modifications and variations of the present invention are possible in light of the above teachings. Therefore, it is to be understood that within the scope of the appended claims, reference numerals are merely for convenience and are not to be in any way limiting, the invention may be practiced otherwise than as specifically described.

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TABLE 1

	ng1	ng2	ng3	ng4	ng5	ng6	ng7	ng8	ng9
[1] gr Pro SCS	1.3	1.3	1.3	1.3	1.3	1.3	1.3	1.3	1.3
[2] gr p 1st dose	1.3	1.3	1.3	1.3	1.3	1.3	1.3	1.3	1.3
	1.3	← check of gr pH							
[3] gr p 2nd dose	1.3	1.3	1.4	1.4	1.4	1.3	1.4	1.3	1.3
	1.3	← check of gr pH							

SCS pH = 8.0.

TABLE 2

Mech Vent	Major Surgery	Multi-trauma	Head Injury	Hypotension	Renal Failure	Sepsis	Multiple Operation	Acid/Base	Coma	Liver Failure	Burn
75	61	35	16	14	14	14	12	10	4	2	2

Risk factors present in patients in this study (n=75)

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TABLE 3

<b>RANITIDINE (day 1 - 9)</b>		
Ranitidine	150mg/24 hr	<u>Per day</u>
Ancillary Product (1)	Piggyback (60%)	6.15
Ancillary Product (2)	micro tubing (etc.)	0.75
Ancillary Product (3)	filter	2.00
Sterile Prep required	yes	.40
R.N. time (\$24/hr)	20 minutes/day (includes pH monitoring)	
R.Ph. time, hood maint.	3 minutes (\$40/hr)	8.00
Pump cost	\$29/24 hrs x 50%	2.00
TOTAL for 9 days		14.50
Note: Does not include the cost of failure and/or adverse effect.		304.20
<b>RANITIDINE Cost per day</b>		<b>33.80</b>
<b>CIMETIDINE (day 1 - 9)</b>		
Cimetidine	900 mg/24 hr	<u>Per day</u>
Ancillary Product (1)	Piggyback	3.96
Ancillary Product (2)	micro tubing (etc.)	1.25
Ancillary Product (3)	filter	2.00
Sterile Prep required	yes	.40
R.N. time (\$24/hr)	20 minutes/day (includes pH monitoring)	
R.Ph. time, hood maint.	3 minutes (\$40/hr)	8.00
Pump cost	\$29/24 hrs x 50%	2.00
TOTAL for 9 days		14.50
Note: Does not include the cost of failure and/or adverse effect.		288.99
<b>CIMETIDINE Cost per day</b>		<b>32.11</b>
<b>SUCRALFATE (day 1 - 9)</b>		
Sucralfate	1 Gm x 4	<u>Per day</u>
Ancillary Product (1)	syringe	2.40
Sterile Prep required	no	.20
R.N. time (\$24/hr)	30 minutes/day (includes pH monitoring)	
TOTAL for 9 days		12.00
Note: Does not include the cost of failure and/or adverse effect.		131.40
<b>SUCRALFATE Cost per day</b>		<b>14.60</b>

Acquisition, preparation and delivery costs of traditional agents.

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TABLE 4

The average length of treatment was 9 days. Cost of care was calculated from these data:

		Per day	Total
<b>OMEPRAZOLE (day 1)</b>			
Product acquisition cost	40 mg load x 2 (5.66/dose)	11.32	11.32
Ancillary product	materials for solution preparation	0.41	0.41
Ancillary product	syringe w/needle	0.20	0.40
Sterile preparation required	no		
SOS preparation time (R.N.)	6 minutes	2.40	4.80
R.N. time (\$24/hr)	21 minutes/day (includes pH monitoring)	8.40	8.40
<b>OMEPRAZOLE (days 2 - 9)</b>			
Product acquisition cost	20 mg per day	2.83	22.65
Ancillary product	materials for solution preparation	0.41	0.82
Ancillary product	syringe w/needle	0.20	1.60
Sterile preparation required	no		
SOS preparation time (R.N.)	6 minutes	2.40	4.80
R.N. time (\$24/hr)	18 minutes/day (includes pH monitoring)	8.00	57.60
2/75 patient require 40 mg simplified omeprazole solution per day (days 2-9)			0.63
No additional cost for adverse effects or for failure			
<b>TOTAL</b>			<b>113.43</b>
<b>Simplified Omeprazole Solution Cost per day</b>			<b>12.60</b>
Pharmacoeconomic evaluation of omeprazole cost of care			

TABLE 5

Time	Control	1 hour	24 hour	2 day	7 day	14 day
Conc (mg/ml)	2.01	2.07	1.94	1.96	1.97	1.98

Stability of Simplified Omeprazole Solution at room temperature (25°C)  
Values are the mean of three samples

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**Claims**

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## CLAIMS

1. A liquid oral pharmaceutical composition,  
comprising:

a) a proton pump inhibitor; and

b) at least one bicarbonate salt of a Group IA metal; wherein if said inhibitor is omeprazole, it must be present in a concentration greater than 1.2 mg/ml, and if said inhibitor is lansoprazole, it must be present in a concentration greater than 0.3 mg/ml.

2. The liquid oral pharmaceutical composition as recited in Claim 1 further comprising a parietal cell activator.

3. The liquid oral pharmaceutical composition as recited in Claim 2 wherein said activator is selected from the group consisting of chocolate, sodium bicarbonate, and calcium and salts, and mixtures thereof.

4. The liquid oral pharmaceutical composition as recited in Claim 1 further comprising a flavoring agent.

5. A solid oral pharmaceutical composition,  
comprising:

a) A proton pump inhibitor; and

b) At least one bicarbonate salt of a Group IA metal; wherein said composition is in a dosage form selected from the group

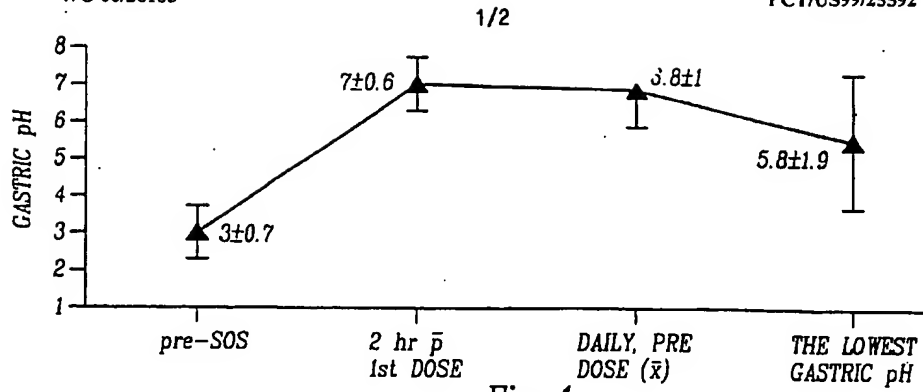
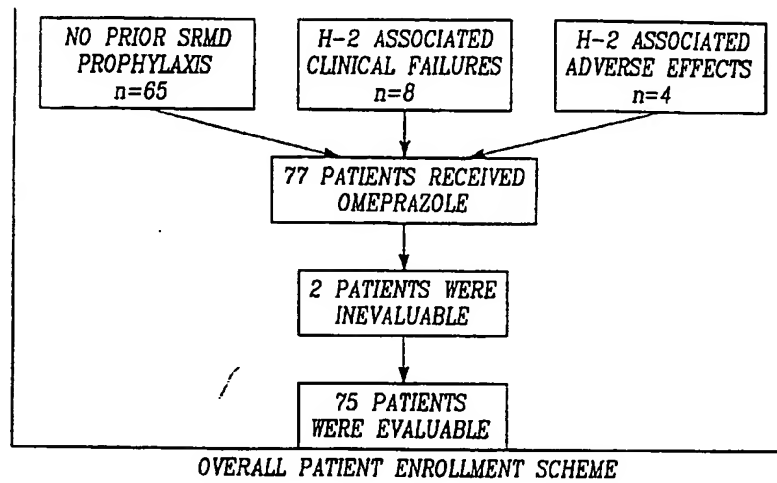
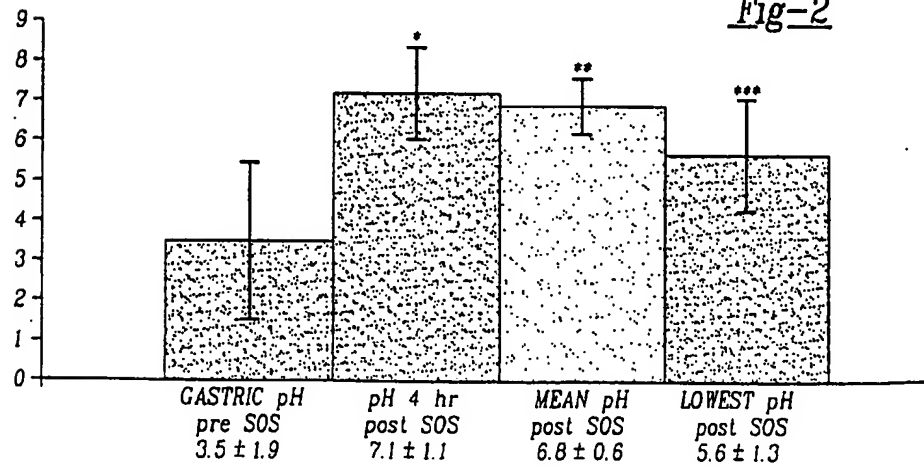
- 57 -

consisting of a powder, a tablet, a capsule,  
an effervescent powder, an effervescent  
tablet, pellets and granules, and wherein  
said dosage form is not enteric coated or  
time-released.

6. The solid oral pharmaceutical composition, as  
recited in Claim 5 further comprising a parietal cell  
activator.

7. A method of treating gastric acid disorders  
comprising administering to a patient an oral  
pharmaceutical composition comprising a proton pump  
inhibitor and at least one bicarbonate salt of a Group IA  
metal wherein said administering step comprises providing  
a patient with a single dose of the pharmaceutical  
composition without requiring further administration of  
the at least one buffering agent.

8. An oral pharmaceutical to be administered in  
combination with a proton pump inhibitor comprising at  
least one bicarbonate salt of a Group IA metal wherein  
said composition is in a dosage form selected from the  
group consisting of a powder, a tablet, a capsule, an  
effervescent powder, an effervescent tablet, pellets and  
granules, and wherein said dosage form is not enteric  
coated or time-released.

Fig-1Fig-2Fig-3

# Chocolate lansoprazole powder vs lansoprazole alone

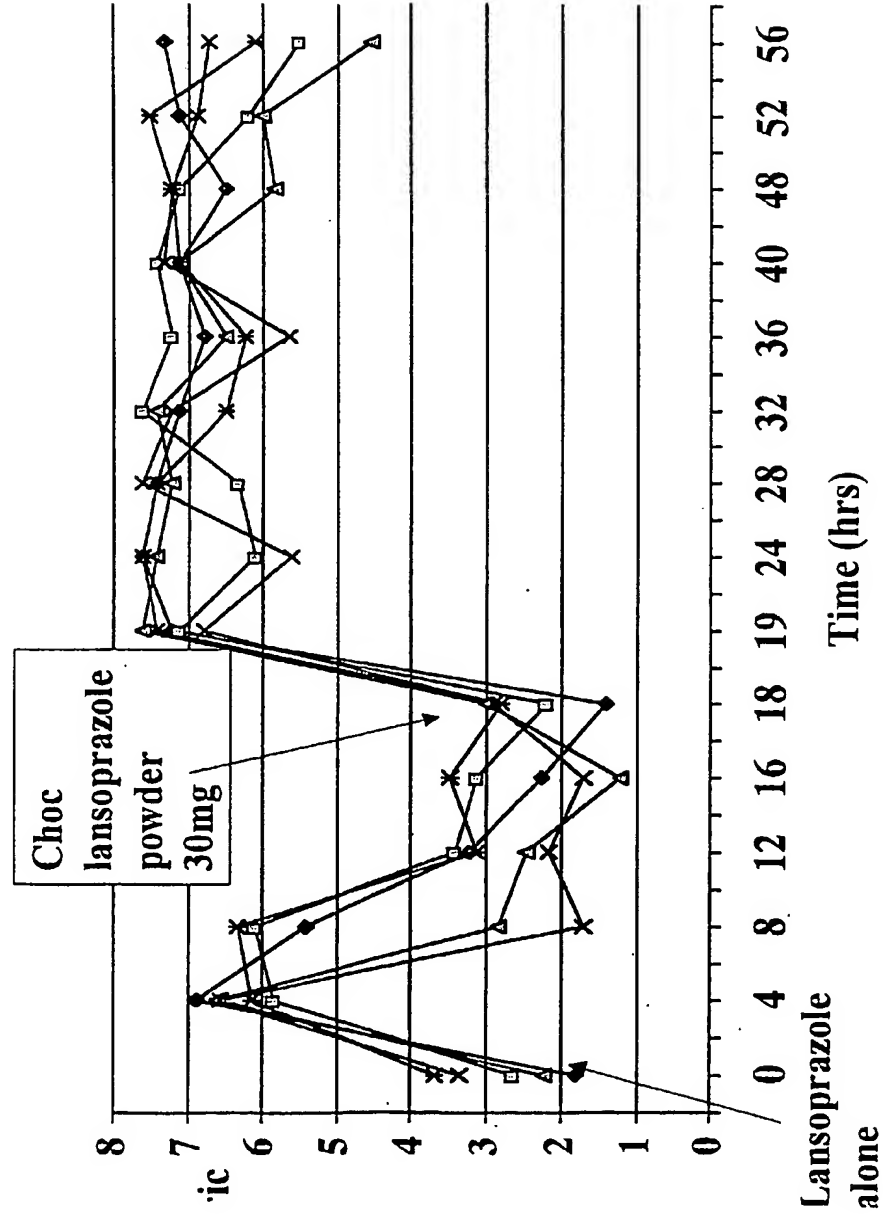


FIGURE 4